

REVISION SHEETS

MEASUREMENT

PHYSICAL QUANTITIES

“The quantities that can be measured and laws of physics are expressed in terms of these quantities are known as physical quantities.”

→ Physical Quantities are divided into :

BASE QUANTITIES

DEFINITION

“Those quantities which cannot be expressed in terms of other quantities are called base quantities”

NUMBER

They are **seven** in number.

EXAMPLES

- length
- Mass
- current
- Temperature
- Intensity of light
- Amount of substance

DERIVED QUANTITIES

DEFINITION

“Those quantities which are expressed in terms of base quantities”

NUMBER

Derived quantities are **infinite**

EXAMPLES

- Area
- Volume
- Force
- Power
- velocity
- Speed.

SYSTEM OF UNITS

MKS \Rightarrow meter - kilogram - second
CGS \Rightarrow centimeter - gram - second
FPS \Rightarrow foot - pound - second

SI UNITS

“The international system of units is a scientific method of expressing the magnitudes or quantities of important natural phenomena”

BASE UNITS

DEFINITION

The units containing base quantities are called base units.

EXAMPLES

meter m
kilogram kg
second s
kelvin K
mole mol
ampere A
candela cd

DERIVED UNITS

DEFINITION

The units which contain derived quantities are called derived units.

EXAMPLES

Area m^2
Volume m^3
acceleration ms^{-2}
speed ms^{-1}
pressure $kgm^{-1}s^{-2}$

SUPPLEMENTARY UNITS

DEFINITION

The units which neither contain base units nor derived units. They contain geometrical units or angles.

EXAMPLES

Radian rad
steradian sr

RADIAN

STATEMENT

"one radian is the angle subtended at the center of a circle by an arc with a length equal to the radius of the circle"

MATHEMATICAL EXPLANATION

$$\text{No. of radians} = \frac{\text{arc length}}{\text{Radius}} = \frac{S}{r}$$

$$\text{one revolution} = 360^\circ$$

$$\begin{aligned} \text{No. of radian in one revolution} &= \frac{2\pi r}{r} \\ &= 2\pi \\ &= 2(3.14) \end{aligned}$$

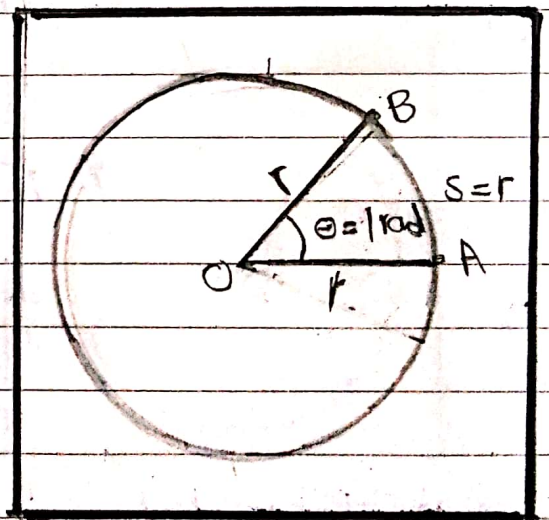
$$\text{No. of radian in one revolution} = 6.28 \text{ rad}$$

$$2\pi \text{ rad} = 360^\circ$$

$$1 \text{ rad} = \frac{360}{2\pi}$$

$$1 \text{ rad} = \frac{360}{2(3.14)}$$

$$1 \text{ rad} = 57.3^\circ$$

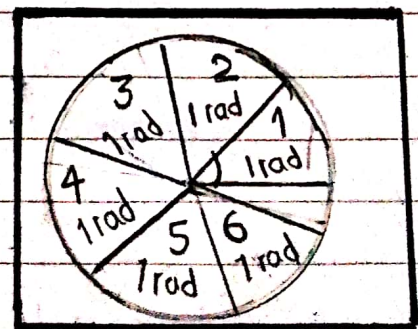


$$1^\circ = \frac{1 \text{ rad}}{57.3}$$

$$1^\circ = 0.017 \text{ rad}$$

* Radian is plane angle

* It is 2 dimensional (2D)



STERADIAN

STATEMENT

“ steradian is the solid angle subtended at the center of sphere by an area of its surface equal to the radius of that sphere. ”

MATHEMATICAL EXPLANATION

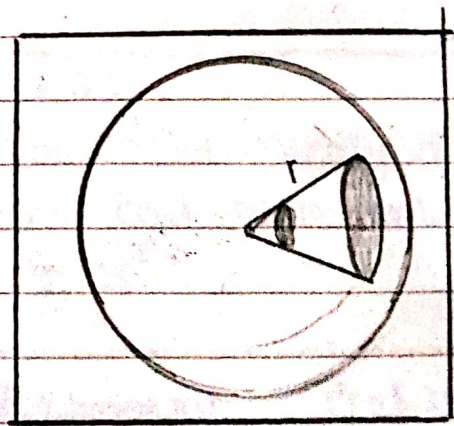
$$\begin{aligned} \text{No. of steradian in sphere} &= \frac{\text{Area of sphere}}{\text{Square of radius}} \\ &= \frac{4\pi r^2}{r^2} \end{aligned}$$

$$\begin{aligned} &= 4\pi \\ &= 4(3.14) \end{aligned}$$

No. of steradian in sphere	= 12.56 sr
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* Steradian is a solid angle

* It is three dimensional (3D)



SCIENTIFIC NOTATION

“ Writing a number in a powers of ten or in a standard form is called scientific notation ”

EXAMPLES

1) $4500000 \Rightarrow 4.5 \times 10^6$

2) $0.00453 \Rightarrow 4.53 \times 10^{-3}$

PREFIXES

“ A mechanism through which a term in scientific notation is expressed by giving a proper name to its power of ten is called prefix. ”

EXAMPLES

$5 \times 10^{-3} \text{ m} \Rightarrow 5 \text{ mm}$

$6 \times 10^9 \Rightarrow 6 \text{ G}$

WAY TO REMBER PREFIXES :

“ Young Zoey Earns Pennies That's Got Mighty King Henry's Daughter Beth • Beth Drinks cold Milk until Nine Pm for A Zillion Years ”

Exponent	Prefix	Exponent	Prefix
24 10^{24}	Yotta	0 10^0	Base
21 10^{21}	Zetta	-1 10^{-1}	deci
18 10^{18}	Exa	-2 10^{-2}	centi
15 10^{15}	Peta	-3 10^{-3}	milli
12 10^{12}	Tera	-6 10^{-6}	micro
9 10^9	Giga	-9 10^{-9}	nano
6 10^6	Mega	-12 10^{-12}	pico
3 10^3	kilo	-15 10^{-15}	femto
2 10^2	hecto	-18 10^{-18}	atto
1 10^1	deca	-21 10^{-21}	zepto
0 10^0	Base	-24 10^{-24}	yocto

Non-SI UNITS

Light Year	The distance that light travels in a vacuum in one year is called Light year. $1 \text{ light year} = 9.4607 \times 10^{15} \text{ m}$
Angstrom	type of prefix that measure small length. $1 \text{ \AA} = 1 \times 10^{-10} \text{ m}$ OR 0.1 nm
Micron	obsolete name of μm which is decimal fraction of the meter. $1 \text{ Micron} = 1 \times 10^{-6} \text{ m}$

WRITING UNIT SYMBOLS

- Unit symbols are printed in roman type e.g.
 - $\text{m} \Rightarrow \text{meter}$ $\text{s} \Rightarrow \text{second}$ $\text{Pa} \Rightarrow \text{pascal}$
- A prefix is never used in isolation and in a compound form e.g.
 - nm pm
 - m μm $\mu\mu\text{m}$
- Multiplication must be indicated by space or dot e.g. Nm OR N.m
 - Division is indicated by horizontal line, by '/' or by -ive exponents.
e.g. m/s OR ms^{-1}

4. Abbreviations for unit symbols or unit names are not allowed. e.g.,

- sec ~~for~~ second OR s
- Sq. mm ~~for~~ mm² OR square millimeter
- cc ~~for~~ cm³ OR cubic centimeter
- mps ~~for~~ ms⁻¹ OR meter per second

5. When multiple of unit is raised to the power, it applies to the whole ~~multiple~~ not just the unit e.g.

cm³ → for whole cm not just m.

WRITING UNIT NAMES

1. Names of the units starts with a lower case letter e.g. joule, hertz, meter

2. The symbol of units named after a scientist has initial capital e.g. 'A' for ampere.

3. Unit can be written in symbols or spelled out in full e.g. 2.1 ms⁻¹ OR 2.1 meter per second.

4. For writing prefix, there should not be space or hyphen b/w prefix name and unit name. e.g.

milligram

milli-gram

ERRORS

“ Error is the doubt that exist about the result of any instrument. For every measurement there is always a margin of doubt which is called error ”

TYPES OF ERRORS

SYSTEMATIC ERRORS

⇒ Those errors that tend to be in one direction (tive or -ive)

INSTRUMENTAL ERRORS CAUSE

- Due to wrong calibration
- Due to L.C of instrument

REDUCTION

1. By changing the instrument
2. By applying correction factor

$$C.F = \text{measured value} \pm Z.E$$

PERSONAL ERRORS CAUSE

- Inexperience of a person
- Due to carelessness
- wrong way of taking reading

REDUCTION

1. Doing work with care
2. advanced, experienced person

RANDOM ERRORS

⇒ Those errors which occur irregularly and random with respect to size and sign

CAUSE

unknown

REDUCTION

By taking several readings and then having their mean

LEAST COUNT

"The smallest value that can be measured by measuring instrument"

EXAMPLE

Least count of meter rule = 0.1 cm

Least count of vernier calliper = 0.01 cm

Least count of screw gauge = 0.001 cm

LEAST COUNT ERROR

"Error associated with least count is called least count error"

METHODS TO REDUCE LEAST COUNT ERROR

1. By using instrument with smaller least count (higher resolution)
2. By improving experimental techniques.

⇒ Least Count Error belongs to random errors.

UNCERTAINTIES

DEFINITION	The amount of possible error associated with a measuring instrument is called uncertainty.
PURPOSE	Uncertainty estimate how small or large the error is. $\text{Physical Quantity} = \text{Observed value} \pm \frac{\text{Least Count}}$
CAUSES	<ol style="list-style-type: none"> 1. Imperfection of person's senses 2. Limitation of measuring instrument 3. Natural variation in objects

TYPES OF UNCERTAINTY

	ABSOLUTE UNCERTAINTY (LEAST COUNT)	RELATIVE/ FRACTIONAL UNCERTAINTY	PERCENTAGE UNCERTAINTY
SYMBOL	It is represented by Δ	It is represented by ϵ	% is written with the value
UNIT	has same unit as the quantity	has no unit	'%' is written with the answer
FORMULA	It is the L.C e.g (0.1 ± 2.1) ⇒ 2.1 is absolute uncertainty To convert % uncertainty to Absolute = $\frac{\% \text{age Un}}{\text{measurement}}$	Relative uncertainty: = $\frac{\text{Absolute uncertainty}}{\text{measurement}}$	%age uncertainty = $\frac{\text{absolute uncertainty}}{\text{measurement}} \times 100$

INDICATING UNCERTAINTIES IN

CALCULATION

⇒ For Sum and difference, "Absolute uncertainties are added"

EXAMPLE

SUM

Let

$$l_1 \pm \Delta l_1 = (6.4 \pm 0.1) \text{ m}$$

$$l_2 \pm \Delta l_2 = (5.3 \pm 0.01) \text{ m}$$

$$\begin{aligned} \text{Sum} &= l_1 + l_2 \\ &= 6.4 + 5.3 \\ &= 11.7 \text{ m} \end{aligned}$$

$$\begin{aligned} \text{Uncertainty} &= \Delta l_1 + \Delta l_2 \\ &= 0.1 + 0.01 \\ &= 0.11 \text{ m} \end{aligned}$$

$$l \pm \Delta l = (11.7 \pm 0.11) \text{ m}$$

DIFFERENCE

$$\text{Let } d_1 \pm \Delta d_1 = (7.4 \pm 0.2) \text{ m}$$

$$d_2 \pm \Delta d_2 = (11.1 \pm 0.2) \text{ m}$$

$$\begin{aligned} \text{difference} &= d_1 + d_2 \\ &= 7.4 + 11.1 \\ &= 18.5 \text{ m} \end{aligned}$$

$$\begin{aligned} \text{Uncertainty} &= \Delta d_1 + \Delta d_2 \\ &= 0.2 + 0.2 \\ &= 0.4 \text{ m} \end{aligned}$$

$$d \pm \Delta d = (18.5 \pm 0.4) \text{ m}$$

⇒ For product and quotient, ^{if} Absolute uncertainties are converted into %age uncertainties then they are added ²²

EXAMPLE

PRODUCT

$$\text{Let } l \pm \Delta l = (1.50 \pm 0.02) \text{ m}$$
$$w \pm \Delta w = (0.20 \pm 0.01) \text{ m}$$

1) Multiply the product

$$A = l \times w$$
$$= 1.50 \times 0.20$$
$$= 0.3$$

2) Convert absolute uncertainty to %age uncertainty

$$\bullet \frac{\Delta l}{l} \times 100$$
$$= \frac{0.02}{1.50} \times 100 \Rightarrow 1.33\%$$

$$\bullet = \frac{0.01}{0.20} \times 100 \Rightarrow 5\%$$

3) Add %age uncertainty

$$= 1.33\% + 5\%$$
$$= 6.33\%$$

4) Convert to absolute uncertainty

$$= \frac{6.33}{100} \times 0.3$$
$$= 0.02$$

$$A \pm \Delta A = (0.3 \pm 0.02) \text{ m}^2$$

QUOTIENT

$$\text{Let } V \pm \Delta V = (5.2 \pm 0.1) \text{ V}$$
$$I \pm \Delta I = (0.84 \pm 0.5) \text{ A}$$

1) Divide the measurement

$$R = \frac{5.2}{0.84}$$
$$= 6.2 \Omega$$

2) Convert absolute uncertainty to %age uncertainty

$$\bullet \frac{0.1}{5.2} \times 100$$
$$= 1.92\%$$

$$\bullet \frac{0.5}{0.84} \times 100 \Rightarrow 59.5$$

3) Add %age uncertainty

$$= 59.5 + 1.92$$
$$= 61.42\%$$

4) Convert back to absolute uncertainty

$$= \frac{61.42}{100} \times 6.2$$
$$= 3.8$$

$$R \pm \Delta R = (6.2 \pm 3.8) \Omega$$

⇒ For Power, %age uncertainty is multiplied with the power²²

EXAMPLE

POWER

Let radius:

$$r \pm \Delta r = (2.25 \pm 0.01) \text{ cm}.$$

For example, we have to find 'area of sphere'

1) Find the volume using formula of area of sphere

$$V = \frac{4}{3} \pi r^3$$

$$= \frac{4}{3} (3.14) (2.25)^3$$

$$= 47.6 \text{ cm}^3$$

2) Convert absolute uncertainty to %age uncertainty

$$\frac{0.01}{2.25} \times 100 \Rightarrow 0.44\%$$

3) Multiply the power in the 'formula' with %age uncertainty

$$3(0.44) \Rightarrow 1.33\%$$

4) Convert back to absolute uncertainty

$$\frac{1.33}{100} \times 47.6 \Rightarrow 0.6$$

$$(47.6 \pm 0.6) \text{ cm}^3$$

SIGNIFICANT FIGURES

Def: In any measurement the accurately known digits and the first doubtful digit are called Significant Figures.

RULES

- 1. NON-ZERO DIGITS** Non zero digits are significant. e.g. 47.872 [5 S.F.]
- 2. ZERO b/w two S. digits** zero b/w 2 S. digits is always significant e.g. 301.5006 [7 S.F.]
- 3. ZEROS TO THE LEFT OF S. digits** zeros to the left of S.F are **not** significant e.g. 0.000 538 [3 S.F.]
- 4. ZEROS TO THE RIGHT OF S.F** zeros to the right of S.F may or may not be significant e.g. 5.200 [4 S.F.]
- 5. SCIENTIFIC NOTATION** In scientific notation OR standard form, figures other than power of ten are significant e.g. 2.1000×10^4 [5 S.F.]

SIGNIFICANT FIGURES IN CALCULATION

Addition And Subtraction

⇒ least decimal place

Multiplication And Division

⇒ least significant figure

EXAMPLES

$$\bullet 246.24 + 238.278 + 98.3 \\ = 582.818 \Rightarrow 582.8$$

$$\bullet 0.35 - 0.1 \\ 0.25 \Rightarrow 0.2$$

$$\bullet 0.9935 \times 10.48 \times 13.4 \\ = 139.519192 \Rightarrow 140$$

$$\bullet 5.5 \div 1.1 \\ = 5 \Rightarrow 5.0$$

PRECISION

- It refers to the degree of exactness
- describe the closeness of set of measurement of same quantity made in same way
- Depends upon the Least Count
- Multiple Readings are required

$$\text{Precision} \propto \frac{1}{\text{Least Count}}$$

Examples

Measurement

Precision

2642 m

1 m

2050 m

10 m

203.05 km

0.01 km

100.050 km

0.001 km

ACCURACY

- describe the closeness of measured value to the actual value.
- does not depend on the Least Count
- A single reading can be accurate

$$\text{Accuracy} \propto \text{Significant Figures}$$

Examples

Measurement

Accuracy

2642

4

2050

3

203.05

5

100.050

6

DIMENSIONS

DEFINITION

Dimensions describe the physical nature of quantity. Each measurable quantity is represented by specific symbol and written within square bracket is called dimension.

SIGNIFICANCE

1. Deriving a formula
2. Check the correctness of equation

DRAWBACK

1. Dimension-less constants can't be obtained
2. can't tell exact relationship b/w quantities in equation
3. does not distinguish b/w quantities having same dimensions.

TERMS USED WITH DIMENSIONS

Dimensional Variables	Dimensional Constants	Dimension-less Variables	Dimension-less constants
D E F I N I T I O N			
The physical quantities which have dimensions of variable magnitude	The physical quantities which have dimensions but are constant in magnitude	The physical quantities which have no dimensions but changing magnitude	The physical quantities which have no dimensions but having constant magnitude.
E X A M P L E S			
<ul style="list-style-type: none">• Force• Energy• Torque• acceleration	<ul style="list-style-type: none">• speed of light• plank's constant• Gravitational constant	<ul style="list-style-type: none">• Radian• steradian• Strain	<ul style="list-style-type: none">• pure numbers• π

mass \Rightarrow m \Rightarrow kg \Rightarrow [M]
length \Rightarrow l \Rightarrow m \Rightarrow [L]
time \Rightarrow t \Rightarrow s \Rightarrow [T]

VELOCITY

$$v = \frac{d}{t}$$

$$v = \frac{m}{s}$$

$$v = \frac{[L]}{[T]}$$

$$v = [LT^{-1}]$$

ACCELERATION

$$a = \frac{v}{t}$$

$$a = \frac{m}{s^2}$$

$$a = \frac{[L]}{[T^2]}$$

$$a = [LT^{-2}]$$

TORQUE

$$\tau = F \times r$$

$$= \text{kgms}^{-2} \times \text{m}$$

$$= \text{kgm}^2\text{s}^{-2}$$

$$= [ML^2T^{-2}]$$

$$S = v_i t + \frac{1}{2} a t^2$$

$$m = \text{ms}^{-1} \cdot s + \frac{1}{2} \cdot \text{ms}^{-2} \cdot s^2$$

$$[L] = m + m$$

$$[L] = 2m$$

$$[L] = [L]$$

FORCE

$$F = ma$$

$$F = \text{kgms}^{-2}$$

$$F = [MLT^{-2}]$$

DENSITY

$$\rho = \frac{m}{V}$$

$$\rho = \frac{\text{kg}}{\text{m}^3}$$

$$\rho = \frac{\text{kg}}{\text{m}^3}$$

$$\rho = [ML^{-3}]$$

WORK

$$W = F \times d$$

$$= \text{kgms}^{-2} \times \text{m}$$

$$= \text{kgm}^2\text{s}^{-2}$$

$$= [ML^2T^{-2}]$$

$$v_f = v_i + at$$

$$\text{ms}^{-1} = \text{ms}^{-1} + \text{ms}^{-2} \cdot s$$

$$[LT^{-1}] = [LT^{-1}] + [LT^{-2}]$$

$$[LT^{-1}] = [LT^{-1}] + [LT^{-1}]$$

$$[LT^{-1}] = 2[LT^{-1}]$$

$$[LT^{-1}] = [LT^{-1}]$$

MOMENTUM

$$p = mv$$

$$p = \text{kgms}^{-1}$$

$$p = [MLT^{-1}]$$

PRESSURE

$$P = \frac{F}{A}$$

$$P = \frac{\text{kgms}^{-2}}{\text{m}^2}$$

$$P = \text{kgms}^{-2}\text{m}^{-2}$$

$$P = \text{kgm}^{-1}\text{s}^{-2}$$

$$P = [ML^{-1}T^{-2}]$$

DISTANCE

$$s = vt$$

$$= \text{ms}^{-1} \times s$$

$$= m$$

$$= [L]$$

$$2as = v_f^2 - v_i^2$$

$$2 \text{ms}^{-2} \cdot m = (\text{ms}^{-1})^2 - (\text{ms}^{-1})^2$$

$$\text{m}^2\text{s}^{-2} = \text{m}^2\text{s}^{-2} - \text{m}^2\text{s}^{-2}$$

$$[L^2T^{-2}] = [L^2T^{-2}] - [L^2T^{-2}]$$

$$[L^2T^{-2}] = [L^2T^{-2}]$$

$$E = mc^2$$

$$= \text{kg} (\text{ms}^{-1})^2$$

$$= \text{kg} \text{m}^2 \text{s}^{-2}$$

$$= [\text{ML}^2 \text{T}^{-2}]$$

$$K.E$$

$$= \frac{1}{2} mv^2$$

$$= \text{kg} (\text{ms}^{-1})^2$$

$$= \text{kg} \text{m}^2 \text{s}^{-2}$$

$$= [\text{ML}^2 \text{T}^{-2}]$$

$$P.E$$

$$= mgh$$

$$= \text{kg} \cdot \text{ms}^{-2} \cdot \text{m}$$

$$= \text{kg} \text{m}^2 \text{s}^{-2}$$

$$= [\text{ML}^2 \text{T}^{-2}]$$

IMPORTANT POINTS

→ All types of energy has same dimension $[\text{ML}^2 \text{T}^{-2}]$

→ All the velocity will have same dimension $v_i, v_f, v_{\text{average}}, v_r$

→ Torque and Work have same dimension but unit of $\vec{\tau}$ is Nm and it is a vector quantity while unit of W is Joule and it is a scalar quantity.
